

## Change in Water Vapor Permeability of Polymer Films Treated with Osmium Tetroxide

Osmium tetroxide staining of polymers containing unsaturated carbon bonds has been used extensively to increase the effective contrast in electron micrographs.<sup>1-3</sup> Sperling et al.<sup>4</sup> have investigated the diffusion-controlled kinetics of the staining process in poly(butadiene-co-styrene)/polystyrene interpenetrating polymer networks. Even polymers which do not contain double bonds stain noticeably with OsO<sub>4</sub>, e.g., the polyurethane-containing interpenetrating polymer networks.<sup>3</sup> In this note we examine the question of whether physical properties other than optical (or electron optical) opacity are affected by OsO<sub>4</sub> staining of saturated, thin (1 mil) commercial polymer films. To investigate this, we examined the water vapor permeability of OsO<sub>4</sub>-treated and untreated films.

We suspended 1-mil-thick films of a commercial copolymer of poly(vinyl chloride)-poly(vinylidene chloride) (trade name Saran) and a polystyrene (trade name Trycite) in the vapor over room-temperature liquid OsO<sub>4</sub> for two days. The treated films were uniformly darkened, and the weight gain of 2.1-cm discs, weighing 0.13 g, of these films after treatment was in the neighborhood of 10<sup>-4</sup> g. The water vapor permeabilities of treated and untreated films were determined using a modified cup technique<sup>5,6</sup> at 22 ± 0.5°C (vapor pressure of water of 2.007 cm Hg). In Table I we list the water vapor permeabilities of the untreated and OsO<sub>4</sub> treated films. In each case an appreciable loss of water vapor permeability resulted from the OsO<sub>4</sub> treatment. The optical darkening of the film appears to be unaffected by the water transmission. The magnitude of the effect is quite large (more than twofold for the polystyrene and almost threefold for the other film). While it is tempting to speculate that this is due to some chemical reaction with monomer, polymer, or film additive which affects the permeability tortuosity factor, we have not been able to demonstrate this experimentally.

TABLE I  
Water Vapor Permeabilities

Film material	Permeability of untreated film, 10 <sup>11</sup> g-cm/cm <sup>2</sup> -cm Hg-sec (average of three runs)	Permeability of OsO <sub>4</sub> -treated film, 10 <sup>11</sup> g-cm/cm <sup>2</sup> -cm Hg-sec (average of three runs)
Poly(vinyl chloride)-poly(vinylidene chloride) (Saran)	0.58	0.21
Polystyrene (Trycite)	18	8.0

This work was supported by the National Science Foundation and the U.S. Army Research Office.

### References

1. D. Pease, *Histological Techniques for Electron Microscopy*, Academic Press, New York, 1965.
2. K. Kato, *Polym. Eng. Sci.*, **9**, 197 (1969).
3. D. Klempner, T. K. Kwei, M. Matsuo, and H. L. Frisch, *Polym. Eng. Sci.*, **10**, 327 (1970); K. C. Frisch, D. Klempner, S. Migdal, H. Ghiradella, and H. L. Frisch, *ibid.*, **14**, 76 (1974); S. C. Kim, D. Klempner, K. C. Frisch, H. L. Frisch, and H. Ghiradella, *ibid.*, **15**, 339 (1975).
4. A. A. Donatelli, D. A. Thomas, and L. H. Sperling, in *Recent Advances in Polymer Blends, Grafts and Blocks*, L. H. Sperling, Ed., Plenum Press, New York, 1974, p. 375.
5. A. C. Newn, *Shirley Institute Memoirs No. 24, J. Text. Inst.*, **41T**, 269 (1950).

6. H. L. Frisch, J. Cifaratti, R. Palma, R. Schwartz, R. Foreman, H. Yoon, D. Klemper, and K. C. Frisch, in *Polymer Alloys*, D. Klemper and K. C. Frisch, Eds., Plenum Press, New York, 1977, p. 97.

HELEN GHIRADELLA

Department of Biological Sciences

J. CIFARETTI  
R. PALMA  
H. L. FRISCH

Department of Chemistry  
State University of New York  
at Albany  
Albany, New York 12222

Received January 30, 1978

Revised June 21, 1978